# Effect of hydroxyapatite loading on protein foamingconsolidation porous alumina

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## **Paper History**

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### **ABSTRACT**

The objective of this work is to investigate the influence of hydroxyapatite (HA) on physical properties of porous alumina bodies fabricated through protein foaming-consolidation method. Hydrothermal derived HA powder was used as bioactive ceramic. Alumina and HA powders were mixed with yolk, starch and darvan 821 A at an adjusted mass ratio to make slurry. HA-toalumina mass ratio was in the range of 0.0 to 0.8 w/w. The slurries were cast into cylindrical shaped molds and then dried for foaming and consolidation process. Subsequently, the dried bodies were burned at 600°C for 1 h, followed by sintering at temperatures of 1,300°C for 2 h. The porous alumina-HA composites with pore size in the range of 50-500 µm and densities of 2.7 - 2.9 g cm<sup>-3</sup> were obtained. Porosity of bodies decreased from 35.9 to 31.7% when HA-to-alumina mass ratio increased from 0.0 to 0.8 w/w. Compressive strength of sintered body was 0.8 MPa at 35.9% porosity and 2.9 MPa at 31.7% porosity. Intensity of tricalcium phosphate (TCP) phase in bodies increased with HA loadings.

**KEY WORDS:** Alumina, Hydroxypatite, Porosity, Protein Foaming-Consolidation, Ceramic.

# 1.0 INTRDODUCTION

Calcium phosphate ceramics have received much attention as potential bone graft substitute because of their biocompatibility, bioactivity and osteoconduction characteristics. The most widely used calcium phosphatebased bioceramics in bone replacement is hydroxyapatite (HA, Ca<sub>10</sub>(PO<sub>4</sub>)<sub>6</sub>(OH)<sub>2</sub>) [1]. HA also has excellent biocompatibility with hard tissues and bioactivity despite its low osteoconductivity and degradation rate, mechanical strength and osteoconductive potential [2]. Porous HA exhibits strong bonding to the bone because the porosity and bioactivity allows the ingrowth of bone tissue to achieve full integration with the living bones. Thus porous HA have been applied in many applications such as for cell loading, drug releasing agents, chromatography analysis and most extensively, for hard tissue scaffolds [3].

Porous HA has low mechanical properties, therefore normally porous HA implants cannot be heavy loaded and are used to fill only small bone defects. This is due to larger pores, strength of the implant decreases significantly [4]. The porous HA are usually very brittle and prone to fracture upon sudden impact, particularly during the healing stage. Therefore, it is desirable to develop scaffold implant materials with both reliable mechanical properties and porous structures, similar or superior to natural bones [5]. To improve the mechanical strength while maintaining the bioactivity of the scaffold, porous alumina-HA have been shown to have higher strength than the HA porous implant and exhibited a similar bioactivity and osteoconductivity property to the HA porous implant [4].

In this report, we report the effect of hydroxyapatite loading on the physical and mechanical properties of porous alumina prepared through protein foaming-consolidation method.

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#### 2.0 METHODOLOGY

The slurries were prepared by dispersing 10 g alumina (Sigma-Aldrich, USA) and hydrothermal derived HA powders with 3 g starch (wheat particles) and 24 g yolk (freshly isolated from chicken egg) as well as dispersant (R.T. Vanderbilt, USA) in a beaker glass. HA loading added into the slurry was 0, 2, 4, 6 and 8 g. The slurries were then mechanically stirred with rate of 150 rpm for 3 h at room temperature. Then the slurries were cast in cylindrical shape mould which has 10.75 mm in diameter and 15.10 mm in height. Subsequently, the slurries were dried in an air oven (Memmerts, 100-800 model) with temperature 180°C for 1 hour. In order to easy mould removal, castor oil was used as lubricant. The dried samples then were burned in a furnace (Protherm, PLF 160/5 model) at a rate of 10°C/min up to 600°C for 1 h for removal of the creating agent and then at rate of 2°C/min up to 1300°C for 2 h.

Electronic densitometer (Alfa Mirage, MD300S model) was used to measure the density of sintered samples. The theoretical density of fully densified alumina, 3.98 g cm<sup>-3</sup> and HA, 3.14 g cm<sup>-3</sup> was used as reference to calculate the total volume fraction of porosity. The crystallinity of samples was analyzed by XRD XRD-6000 (Shimadzu Diffractometer, model). macrostructures of bodies were examined using SEM (JEOL, 5600 model). The mechanical strength of the specimens (typically 15 mm height and 10 mm diameter) was measured using a universal testing machine (Lloyd, LR10K plus model) by diametrical compression at a loading rate of 2.5 mm min<sup>-1</sup>. Five specimens were used to determine the average maximum compressive strength.

# 3.0 RESULTS AND DISCUSSION

The result of measurements for shrinkage, relative density and porosity of porous alumina-HA composite after sintered at 1300°C is described in Table 1.

Table 1: The Effect of Slurry Composition on The Physical Properties

HA-to-	Shrinkage	Density	Porosity	Compressive
alumina mass	(vol.%)	(g/cm <sup>3</sup> )	(%)	Strength
ratio (w/w)				(MPa)
0.0	58.8	2.55	35.9	0.77
0.2	61.0	2.64	33.4	1.30
0.4	66.7	2.65	33.1	1.96
0.6	67.6	2.66	32.9	2.23
0.8	70.5	2.71	31.7	2.93

When the HA-to-alumina mass ratio increased from 0.0 to 0.8 w/w, the shrinkage increased from 63.5 to 78.4 vol.% The melting point of HA is 1650°C and at approximately 1350°C, HA starts to decompose into highly degradable  $\alpha\text{-TCP}$  [6]. When the composite was sintered at 1300°C, the TCP particles melted and resulted in smaller pore sizes and denser walls. Thus, the shrinkage and density increased. The density of sintered sample was found in the range of 2.55 to 2.71 g cm $^3$ . The shrinkage occurred as the yolk and starch particles were removed, thus the shrinkage became intensive with the increasing sintering temperature.

The porosity decreased from 32.7 to 27.6% when the HA-to-alumina mass ratio increased from 0.0 to 0.8 w/w. The

compressive strength of porous materials is strongly affected by the strength of the materials wall (struts) and the surface defects of the strut. Porosity is also considered to have a significant impact on compressive strength. The compressive strength remarkably increased from 0.8 to 2.9 MPa with the decreasing porosity from 35.9 to 31.7% as shown in Figure 1. Besides the nanostructures HA powder can improve the sinterability due to higher surface energy and improves mechanical properties [7].

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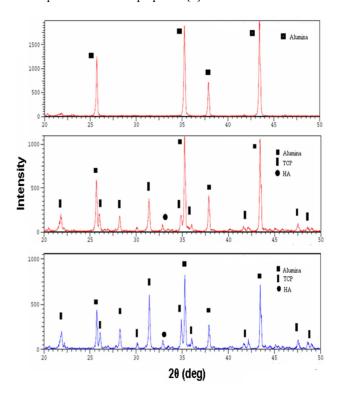


Figure 1: XRD patterns of sample containing HA-to-alumina ratio of (a) 0, (b) 0.4 and (c) 0.8 w/w

Figure 1 shows the result of XRD peaks of porous alumina-HA composites prepared using slurry containing HA-to-alumina mass ratio of 0.0, 0.4 and 0.8 w/w. Sample without any HA loading shown that the alumina peak obtained at 43.4770, 35.2798, 25.7011 and 37.9037  $2\theta$  with intensity of 1051, 917, 544 and 361 cps respectively. As HA loading increase at 4 g, the strongest peak of alumina was obtained at 25.6888, 35.2679, 37.8879 and 43.4627  $2\theta$  at intensity of 255, 549, 194 and 550 cps respectively. The highest TCP peak was occurred at 31.3969  $2\theta$  at intensity of 169 cps. While HA peak was marked at 32.8733  $2\theta$  at

small intensity of 36 cps. As HA loading increase at 8 g, the strongest peak of alumina was obtained at 25.7361, 35.3117, 37.9273 and 43.5085  $2\theta$  at intensity of 203, 397, 129 and 355 cps respectively. The highest TCP peak was occurred at 31.4407  $2\theta$  at intensity of 261 cps. While HA peak was marked at 32.9139  $2\theta$  at small intensity of 42 cps.

At sintering temperature above  $1350^{\circ}$ C, HA becomes unstable and forms decomposed phases such as  $\alpha$ -TCP and  $\beta$ -TCP, tetracalcium phosphate and calcium oxide [8]. It is well known that the solubility of TCP is much higher than that of HA. Moreover there is a small difference among the three patterns observed and no additional phase was identified. This indicates that the sintering process does not change the composition of the porous composite. When HA-to-alumina mass ratio increased from 0.2 to 0.8 w/w, the peaks of HA and TCP become more crystalline but in a small amount of intensity due to HA as the minor composition in this porous composite.

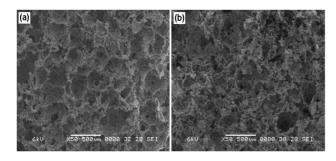


Figure 2: SEM image of bodies with HA-to-alumina mass ratio of (a) 0.0 and (b) 0.8 w/w

The SEM micrograph of sintered porous alumina-HA samples are shown in Figure 2. The bodies have pore size in the range of  $50-500~\mu m$  and completely interconnected. The pore sizes decreased when HA-to-alumina mass ratio increased thus provide a good density and strength of the porous body. The pore size of pure alumina body (sample 1) was largest thus give low in density and low mechanical properties. As the HA-to-alumina mass ratio increased from 0.0 (sample 1) to 0.8~w/w (sample 5), the pores become less interconnected with more dense and thicker pore wall and it is important factors that improve the mechanical properties of the porous sample [9].

## 4.0 CONCLUSION

The effect of HA loading on the physical properties of protein foaming-consolidation porous alumina has reported. The physical properties of the porous sample can be controlled by manipulating the HA-to-alumina mass ratio in the initial slurry. The shrinkage and density of bodies increased from 58.8 to 70.5 vol.% and 2.55 to 2.71 g cm $^3$ , respectively when the HA-to-alumina mass ratio from 0.0 to 0.8 w/w. The compressive strength obtained by this method was 0.8 MPa - 2.9 MPa with porosity between 35.9 – 31.7% and also have open, interconnected porous structure with pore size of 95-300  $\mu m$ . XRD results shows that the cystallinity of TCP phase increased with the increasing HA loading.

#### REFERENCE

- Yusof, M.R., Shamsudin, R., Baharuddin, S. and Besar, I. (2008). Fabrication of porous hydroxyapatite for bone graft substitutes via gas technique. Sains Malaysiana, Vol. 37, pp. 395-399.
- Saki, M., Kazemzadeh, M., Samadikuchaksaraei, A., Basir, H. and Gorjipour, F. (2009). Biocompatibility study of a hydroxyapatite-alumina and silicon carbide composite scaffold for bone tissue engineering. Yakhteh Vol. 11, pp. 55-60.
- Abdurrahman, T. and Sopyan, I. (2008). Recent progress on the development of porous bioactive calcium phosphate for biomedical applications. Recent Patents on Biomedical Engineering Vol. 1, pp. 213-229.
- Youn, K.J., Wan, H.K., Oh, K.K. and Seong, H.H. (2003). The fabrication and biochemical evaluation of alumina reinforced calcium phosphate porous implants. Biomaterials Vol. 24, pp. 3731-3739.
- Hong, T.F., Guo, Z.X. and Yang, R. (2008). Fabrication of porous titanium scaffold materials by a fugitive filler method. Journal of Materials Science: Materials in Medicine Vol. 19, pp. 3489-3495.
- Fadli, A. and Sopyan, I. (2011). Porous ceramics with controllable properties prepared by protein foamingconsolidation method. Journal of Porous Materials Vol. 18, pp. 195-203.
- 7. Banarjee, A., Bandyopadhyay A. and Bose, S. (2007). Hydroxyapatite nanopowders: synthesis, densification and cell-material interactions. Materials Science and Engineering C, Vol. 27, pp. 729-735.
- 8. Muralithran, G. and Ramesh, S. (2000). The effects of sintering temperatures on the properties of hydroxyapatite. Ceramics International Vol. 26, pp. 221-230.
- 9. Ramay, H.R. and Zhang, M. (2003). Preparation of porous hydroxyapatite scaffolds by combination of the gel-casting and polymer sponge method. Biomaterials, Vol. 24, pp. 3293-3302.